

KAPNIK, G.M.; MIKHLIN, S.Ya.; KAPNIK, L.I. (Moskva)

Detection of intestinal function disturbances in infectious hepatitis patients. Vrach.delo no.8:827-829 Ag '59. (MIRA 12:12)

1. Laboratoriya fiziologii pishchevareniya (zav. - prof. G.K. Shlygin)
Instituta pitaniya AMN SSSR i Moskovskaya infektsionnaya klinicheskaya
Bol'ница No.1.
(HEPATITIS, INFECTIOUS) (INTESTINES--DISEASES)

MIKHLIN, S.Ya.

Excretion of enzymes with feces in puppies in certain experimental nutritional disorders; preliminary report. Pediatriia 37 no.10:
34-38 O '59. (MIRA 13:2)

1. Iz laboratorii pishchevareniya (zaveduyushchiy - prof. G.K. Salygin)
otdela fiziologii i biokhimii pitaniya Instituta pitaniya AMN SSSR.
(INFANT NUTRITION DISORDERS exper.)
(ENZYMEs metab.)
(FECES chem.)

MIKHLIN, S.Ya.; PAVLOVA, Z.H.

Effect of aminopterin on the secretion of intestinal enzymes.
Physiol.zhur. 45 no.6:698-704 Je '59. (MIRA 12:8)

1. From the laboratory of physiology of digestion, Institute
of Nutrition, Moscow.

(AMINOPTERIN, eff.

on intestinal enzyme secretion in dogs (Rus))
(INTESTINES, physiol.

enzyme secretion, eff. of aminopterin in dogs
(Rus))

(ENZYMES

intestinal, eff. of aminopterin on secretion
in dogs (Rus))

ALIYEVA, V.I., kand.meditsinskikh nauk; MIKHLIN, S.Ya., kand.biologicheskikh nauk

Changes in the intestinal enzymes in chronic colitis as shown by data on the late results of treatment. Vrach. delo no.8:70-73 Ag '60,
(MIRA 13:9)

1. Laboratoriya fiziologii pishchevareniya (zav. - prof. G.K. Shlygin)
i klinika lechebnogo pitaniya (zav. - prof. F.K. Men'shikov) Instituta
pitaniya AMN SSSR.

(COLITIS)

(DIGESTIVE FERMENTS)

MIKHIIN, S. YA. (USSR)

"The Value of Enzymic Indices in the Diagnosis of Intestinal Disorders."

Report presented at the 5th International Biochemistry Congress,
Moscow, 10-16 Aug 1961

MIKULIN, S.Ya.; NESTERIN, M.F. (Moskva)

Secretory function of the pancreas and intestines in folic acid deficiency caused by the administration of aminopterin. Pat. fiziol. i eksp. terap. 5 no.6:26-30 N-D '61. (MIRA 15:4)

1. Iz laboratorii fiziologii pishchovareniya (zav. - prof. G.K. Shlygin) Instituta pitaniya AMN SSSR.
(PANCREAS—SECRECTIONS) (FOLIC ACID)
(DIGESTIVE ENZYMES) (GLUTAMIC ACID)

MIKHLIN, S.Ya. (Moskva)

Evaluation of the activity of the intestines with the aid of
enzyme indices in diseases of the digestive tract. Klin.med.
no.7:61-67 '61. (MIRA 14:8)

1. Iz laboratorii fiziologii fiziologii pishchevareniya (zav. -
prof. G.K. Shlygin) Instituta pitaniya AMN SSSR.
(ENZYMES) (DIGESTIVE ORGANS—DISEASES)

MIKHLIN, S.YA.

MIKHLIN, S.Ya.; NESTERIN, M.F. (Moskva)

Gastric secretory function during folic insufficiency due to
the administration of aminopterin. Vrach. delo no.8:3-6 Ag '61.
(MIRA 15:3)

1. Laboratoriya fiziologii pishchevareniya (zav. - prof. G.K.
Shlygin), radiobiologicheskaya laboratoriya (zav. - G.P.
Yeremin) Instituta pitaniya AMN SSSR.

(STOMACH--SECRECTIONS)
(FOLIC ACID) (AMINOPTERIN)

YEZOROV, M. N., prof.; MIKHLIN, S. Ya.; TUZHILIN, S. A.

Amount of intestinal enzymes in the duodenal contents and feces in patients with diabetes mellitus. Terap. arkh. 33 no. 5:68-73 My '61.
(MIRA 14:12)

1. Iz otdela lechevnogo pitaniya (zav. - doktor meditsinskikh nauk L. M. Levitskiy) i laboratorii fiziologii pishchevareniya (zav. - prof. G. K. Shlygin) Instituta pitaniya AMN SSER.

(DIABETES) (ENZYMES) (INTESTINES)
(FECES--ANALYSIS)

ALIYEVA, V.I.; MIKHLIN, S.Ya.

Amount of enterokinase and pancreatic enzymes in the duodenal
juice of patients with chronic colitis and enterocolitis. Med.
zhur.Uzb. no.3:38-40 Mr '62. (MIRA 15:12)

1. Is laboratorii fisiologii pishchevareniya (zav. - prof.
G.K.Shlygin) i otdela lechebnogo pitaniya (zav. - doktor med.
nauk L.M.Levitskiy) Instituta pitaniya AMN SSSR.
(COLITIS) (PANCREAS—SECRECTIONS)

MIKHLINE, S. Ya.; NESTEROV, M.F.; BOCHKOV, N.P. (Moskva)

Effect of β -acetylpyridine on the enzyme-excretory processes
and morphological picture of the intestine in dogs. Pat. fiziol.
i eksp. terap. 6 v. 3:67-68 My-Je'62 (MIRA 17:.)

1. Iz laboratorii fiziologii pishchevareniya (zav. - prof.
G.K.Shlygin) Instituta pitaniya AMN SSSR i Instituta eksperi-
mental'noy patologii i terapii AMN SSSR (direktor - doktor
med. nauk L.A. Lapin).

NESTERIN M.F.; MIKHLIN, S.Ya.

Enzyme excretory processes in the principal digestive glands
in some experimental avitaminoses. Vop. pit. 22 no.5:19-22
S-0 '63. (MIRA 17:1)

1. Iz laboratorii fiziologii i patologii pishchevareniya
(zav. - prof. G.K. Shlygin) Institut pitaniya AMN SSSR,
Moskva.

NESTERIN, M.F.; MIKHILIN, S.Ya.

Effect of the deficiency of some group B vitamins on the ~~external~~
secretory activity of the pancreas. Biul. eksp. biol. i med. 57
no.1:41-44 Ja '64. (MIRA 17:10)

1. Laboratoriya fiziologii i patologii pishchevareniya (zav. -
prof. G.K. Shlygin) Instituta pitaniya AMN SSSR. Predstavlena
deystvitel'nym chlenom AMN SSSR B.A. Lavrovym.

ALIYEVA, V.I.; MIKHLIN, S.Ya.

Change in some enzymatic indices in chronic intestinal diseases.
Sov. med. 28 no.7:28-34 Jl '64. (MIRA 18:8)

1. Laboratoriya fiziologii i patologii pishchevareniya (zav. -
prof. G.K.Shlygin) i otdel lecheniya pitaniya (zav. - doktor
med.nauk I.S.Savoshchenko) Instituta pitaniya AMN SSSR, Moskva.

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CIA-RDP86-00513R001134120016-2

APPROVED FOR RELEASE: 06/14/2000

CIA-RDP86-00513R001134120016-2"

L 15740-66

ACC NR: AP5027346 .(N)

SOURCE CODE: UR/0396/65/009/005/0032/0036

AUTHOR: Nesterin, M. F.Mikhlin, S. Ya.24
B

ORG: Laboratory of Physiology and Pathology of Digestion, Institute of Nutrition, ANN SSSR, Moscow (Laboratoriya fiziologii i patologii pishchevareniya Instituta pitaniya ANN SSSR)

TITLE: Changes in the activity of digestive organs during administration of β -acetylpyridine

SOURCE: Patologicheskaya fisiologiya i eksperimental'naya terapiya, v. 9, no. 5, 1965, 32-26

TOPIC TAGS: drug effect, dog, rat, gland, digestion, digestive system, digestive drug, morphology

ABSTRACT: Beta-acetylpyridine was used as an antagonist to investigate the influence of nicotinic acid on the functions of organs and tissues of the alimentary canal. The administration of 192 mg of β -acetylpyridine to a dog with nicotinic

1/2

UDC: 612.3.014.46:615.756.1

L 15740-66

ACC NR: AF5027346

acid deficiency led to its rapid death. The addition of β -acetylpyridine to the food of rats produced nicotinic-acid deficiency symptoms, subsequently prevented by the administration of nicotinic acid, nicotinamide, or tryptophane. The nicotinic-acid deficiency caused by the administration of a specific anti-metabolite caused a nutritional deficiency without structural alterations in the glandular cells. When β -acetylpyridine was given parenterally to healthy dogs in repeated doses of 1.0-2.0g/kg caused the development of a nicotinic-acid deficiency accompanied by changes in the secretory activity in the gastric and pancreatic glands, and by altered enzyme-secretory functions in the intestinal glands. The morphological structure of the oral mucosa, liver, and large and small intestines also changed. Orig. art. has 2 figures and 3 tables.

SUB CODE: 06/ SUBM DATE: 17Mar64/ NR REF Sov: 006/ OTHER: 007

2/2

MIKHLIN, S.Ya. (Moskva)

Effect of experimental folic acid deficiency, caused by the administration of aminopterin, on the secretion of enzymes by the main digestive glands. Arkh. pat. 24 no.9:24-32 '62.
(MIRA 17:4)

1. Iz laboratorii fiziologii pishchevareniya (zav. - prof. G.K. Shlygin) Instituta pitaniya AMN SSSR.

MUKHLIN, S.Ya.; NESTERIN, M.F. (Mnokva)

Secretory function of the stomach in zinc deficiency induced by
 β -acetylpyridine administration. Part first. Ieksp. form., 7
no.6161-64 N-3 '63.
(MIR) 1971

1. Iz laboratori fiziol. i pishchevareniya (zav. prof.
G.M. Shlygin) Instituta pitanija MN SSSR.

ACCESSION NR: AR4042249

S/0081/64/000/008/S020/S020

SOURCE: Ref. zh. Khimiya, Abs. 8S102

AUTHOR: Rayevskiy, V. G.; Yegorov, Ye. V.; Mikhlin, V. E.; Gul', V. Ye.;
Voyutskiy, S. S.

TITLE: Influence of radiochemical cross-linking of elastomers on their adhesion
to fiberforming polymers

CITED SOURCE: Sb. Vy'sokomolekul. soyedineniya. Adgeziya polimerov. M., AN SSSR,
1963, 89-93

TOPIC TAGS: elastomer, adhesion, polymer, radiochemistry, radiation vulcanization

TRANSLATION: The change of durability of adhesion of elastomers SKS-30 ARM-15, SKN-26
and butyl rubber with polycaprolactam film during irradiation of samples by a flow
of accelerated electrons was examined. It was determined that the change of
resistance to separation during irradiation is described by curves passing
through a maximum which corresponds to a definite integral dose of irradiation.

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ACCESSION NR: AR4042249

Thus the character of the change in adhesion strength during radiation vulcanization does not differ qualitatively from that observed earlier for cases of thermal vulcanization in the presence of vulcanizing agents. For samples with coatings of SKS-30 ARM-15 the dependence of the adhesion of this elastomer to polycaprolactam film was studied from the degree of its cross-linking during irradiation. The latter was characterized by the length of the section of molecular chain (M_c), included between two nodes of the space lattice. It was shown that the limiting degree of cross-linking, after the achievement of which a drop of adhesion strength sets in, shifts under the influence of radiation in the direction of a smaller density of the lattice, as compared to that observed for thermal vulcanization in the presence of vulcanizing agents. This phenomenon is explained from the positions of diffusion theory of adhesion. The presence of a limiting degree of cross-linking during radiation vulcanization was observed also on rubber-fabric materials based on capron fabric with a coating of Nairit and SKS-30 ARM-15 applied by facing the fabric on a calender. From authors' abstract.

SUB CODE: MT, OC

ENCL: 00

Card 2/2

I-40010-65 EWD(j)/EWT(m)/EPP(c)/EPP(n)-2/EWD(v)/ENP(v)/EPB/ENP(j)/T/EWA(b)/
EWA(1) Fe-4/Pt-5/Pr-4/Ps-4/Pu-4/Peb RPL GO/RM/WW/03
ACCESSION NR: AT4049836 S/0006/64/000/000/0008/0012

56

52

BT

AUTHOR: Gol'danskiy, V. I.; Gul', V. Ye.; Yegorov, Ye. V.; Zil'berg, G. A.;
~~Mikhlin, V. E.~~; Rayevskiy, V. G.

TITLE: A new radiochemical method for preparing graft copolymers, and their
possible uses for increasing the bond strength between rubber and fabric

SOURCE: Khimicheskiye svoystva i modifikatsiya polimerov (Chemical properties
and the modification of polymers); sbornik statey. Moscow, Izd-vo Nauka, 1964,
8-12

TOPIC TAGS: graft copolymer, bond strength, rubber fabric laminate, neutron
irradiation, polycaproamide, elastomer, polymer impregnation, Capron fabric

ABSTRACT: Utilizing the localized effect of neutron irradiation, a new method
was developed for obtaining graft copolymers; this was based on the irradiation
of emulsions containing both polymer components and a lithium (boron) compound by
a flow of thermal neutrons. The graft copolymers tested were obtained by irradi-
ation, in a nuclear reactor, of emulsions made from a mixture of polycaproamide
in formic acid, containing a Li compound, with solutions of elastomers in o-xylene.
Infrared spectra showed the presence of a radiochemical interaction between the
elastomer molecules and polycaproamide with the formation of a graft copolymer.

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ACCESSION NR: AT4049836

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The composition of the resin mixture is tabulated. The resin coating was 0.2-0.02 mm thick. The vulcanized samples were tested on a Schopper apparatus. Tabulated data show that impregnation of Capron¹² fabric with a non-irradiated emulsion decreases the bond strength between rubber and fabric by 30-40%, due to a decrease in the mechanical adhesion and the low cohesive strength of the adhesive. The use of the impregnating solution containing graft copolymer increases the bond strength by 45-60% as compared to the initial value. By combining impregnation of the fabric with a solution of epoxyamide resin (No. 89) and impregnation with a solution of an elastomer and a graft copolymer, the bond strength between the rubber and the fabric was almost doubled as compared to the strength obtained by impregnating only with epoxyamide, and increased four times as compared to materials based on nonimpregnated Capron fabric. Other modifications of the method of localized neutron irradiation permit the bond strength to be increased to 4.1 kg/cm², this value being limited by the cohesion of the rubber coating.¹⁵ This variant of the method will be described in a subsequent publication. Orig. art. has: 1 figure and 3 tables.

ASSOCIATION: Institut khimicheskoy fiziki AN SSSR (Chemical physics institute, AN SSSR); Moskovskiy institut tonkoy khimicheskoy tekhnologii im. M. V. Lomonosova (Moscow fine chemical technology institute)

SUBMITTED: 18Apr62 ENCL: 00 SUB CODE: OC, MT
Cord 2/2 NO REF SOV: 006 OTHER: 001

ACC NR: AP7001406

SOURCE CODE: UR/0413/66/000/021/0108/0109

INVENTOR: Dontsov, A. A.; Mikhlin, V. E.; Dogadkin, B. A.

ORG: none

TITLE: A method for vulcanizing synthetic rubbers. Class 39, No. 187995 (announced by Moscow Institute of Fine Chemical Technology im. M. V. Lomonosov (Moskovskiy institut tonkoy khimicheskoy technologii))

SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 21, 1966,
108-109

TOPIC TAGS: synthetic rubber, ~~synthetic rubber~~ vulcanization, carboxylate acid,
monovalent metal carboxylate

ABSTRACT: An Author Certificate has been issued for a method for vulcanizing synthetic rubbers with organic acid derivatives in the presence of radical-type activators. To improve the technological properties of rubber mixtures, salts of monovalent metals of unsaturated carboxylic acids are used as the organic acid [BO]

SUB CODE: 11, 07/ SUBM DATE: 16Jul65/ XTD PRESS: 5109

Card 1/1

UDC: 678.7.028.294:547.39-38

SOSONKIN, Z.S.; MIKHLIN, V.M. (Kishinev)

Chronic subdural hematoma. Vop.neirokhir. no.2:49-53 '62.
(MIR 15:3)

1. Neyrokhirurgicheskoye i rentgenologicheskoye otdeleniya
Respublikanskoy psichonevrologicheskoy bol'nitsy Moldavskoy
SSR.

(HEMATOMA) (DURA MATER - TUMORS)

MIKHLIN, V.M.

Reorganization and functional adaptation of the cranial bones as
a sequel of meningoencephalitis in childhood. Zdravookhranenie
5 no.3:41-44 My-Je '62. (MIRA 16:1)

1. Iz kafedry psichiatrii (zav. - prof. A.N.Molokov) Kishinev-
skogo meditsinskogo instituta.
(SKULL—ABNORMITIES AND DEFORMITES) (ENCEPHALITIS)

MIKHILIN, V.M. (Kishinev)

Acute stomach dilatation in a patient with thyrotoxicosis.
Probl. endok. i gorm. 9 no. 5 1976-78 S-0'63 (MIRA 16:12)

1. Iz Respublikanskoy psichoneurologicheskoy bol'nitsy (glavnyy
vrach B.A. Morozov) Moldavskoy SSR.

MIKHLIN, V. M. Cand Tech Sci -- "Study of the ~~process of the~~ clogging ~~process of the~~ consistent lubrication ~~etc.~~ and certain peculiarities of their behavior in bearing units of tractors and agricultural machines." Mos, 1960 (Min of Agr RoFSR. Len Agr Inst.).
(KL, 1-01, 195)

-216-

MIKHLIN, V.M.

Conference of an international study group on problems concerning
the repair of agricultural machinery. Mekh. i elek. sots. sel'khoz.
20 no.1:64 '62. (MIRA 15:2)
(Agricultural machinery--Repairing)

18.4100

77940
SCY 47-87-11-1-1

AUTHORS: Mikhlin, V. M., Speransov, N. N.

TITLE: Determining Abrasive Impurities in Petroleum Products

PERIODICAL: Khimiya i tekhnologiya topliv i masel, 1960, Nr 3,
pp 57-61 (USSR)

ABSTRACT: The described method uses two plates, one of blued steel and the other of cardboard, between which a sample of the petroleum product is placed. When the plates are rubbed against each other the abrasive impurities in a given petroleum product remove the dark oxide film from the metal plate. Since the polished (light) area of the metal plate is proportional to the amount of the abrasive admixtures, one can plot a calibrated curve and determine the concentration of the abrasive admixtures in petroleum products. The best results are obtained when the test is made at 18-20°. There are 4 figures; 1 table; and 2 references, 1 Soviet, 1 U.S. The U.S. reference is Zweifel, H. C., ASTM Symposium, Nr 34, 20 (1948).

ASSOCIATION: All-Union Scientific Research Institute of Mechanization of Agriculture (VIM) Card 1/1

MIKHLIN, V.M.

Efficient lubrication and prolongation of the life of antifriction bearings. Trakt. i sel'khozmash. 30 no.8:30-32 Ag '60.
(MIRA 13:8)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut mekhanizatsii
sel'skogo khozyaystva.
(Bearings (Machinery))

S/272/63/000/001/002/006
E194/E155

AUTHOR: Mikhlin, V.M.

TITLE: A new method of determining abrasive contaminants in petroleum products

PERIODICAL: Referativnyy zhurnal, otdel'nyy vypusk, Metrologiya i izmeritel'naya tekhnika, no.1, 1963, 42-43, abstract 1.32.303. (In collection: Metody, pribory i oborud., primenayemyye pri issled. i ispytaniis.-kh. tekhn. (Methods, instruments and equipment used in agricultural engineering investigations and testing). M., 1961, 268-277)

TEXT: The Vsesoyuznyy nauchno-issledovatel'skiy institut mekhanizatsii sel'skogo khozyaystva (All-Union Scientific Research Institute for Mechanization of Agriculture) has developed a method of assessing the abrasive properties of mechanical contaminants in petroleum products. The sample of oil is spread between plates, one being of metal with a dark oxidized surface and the other of a material which is able to retain hard cardboard particles on its surface. During spreading, the abrasive contaminants scratch the surface.

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S/272/63/000/001/002/006
E194/E155

A new method of determining ...

metal plate and brighten it in the contact zone. The device for spreading the oil consists of a load holder, a frame and a base-plate. The holder presses with constant force on the inserted metal plate which is bent to a certain profile to set up a definite contact. The frame, which is moved by a crank, is slotted to take a sheet of cardboard which carries the petroleum product to be tested. The size of the instrument is 40 x 40 x 105 mm; its weight 0.7 kg. The test lasts 3 - 5 minutes. The size of the light band on the metal plate is measured visually for the qualitative analysis of oil products or with a photo-electric calorimeter type ФЭК-М (FEK-M) suitably adapted for accurate quantitative work. The test procedure, the method of preparing the metal sheets and the use of control samples are described. The results of analysis of oil products with various amounts of quartz contaminant establish curves relating the degree of wear of plates to the concentration of abrasive contaminants for diesel fuel, diesel lubricant ДП-8 (Dp-8) and grease УСС-2 (USS-2). The tests responded to abrasive concentrations of the order of 0.002%. The influence of the hardness of contaminant particles on sheet wear and the influence of the oil sample size on the test results

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A new method of determining ...

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E194/E155

are shown. A schematic diagram of an automatic device for determining abrasive contaminants has been developed so that a large number of analyses can be carried out.
6 figures.

Abstractor's note: Complete translation.]

Card 3/3

PASECHNIKOV, N., inzh.; MIKHLIN, V., inzh.

M
Position of the release valve in coarse oil filters.
Trakt. i sel'khozmash. 31 no.7:42 Jl '61. (MIRA 14:6)

1. Vsesoyuzny nauchno-issledovatel'skiy institut mekhanizatsii
sel'skogo khozyaystva.
(Diesel engines--Oil filters)

MIKHLIN, V.M., kand. tekhn. nauk

Evaluating the repairing qualities of machine parts. Mekh. i
elek. sots. sel'khoz. 21 no.1:24-26 '63. (MIRA 16:7)

1. TSelinnyy filial Gosudarstvennogo vsesoyuznogo nauchno-
issledovatel'skogo tekhnologicheskogo instituta remonta i
ekspluatatsii mashinno-traktornogo parka.
(Agricultural machinery—Maintenance and repair)

L-11865-66 EWT(m)/EWP(t)/EWP(b) IJP(c) JD/JG

ACC NR: AP6000763

UR/0078/65/010/012/2787/2795

AUTHOR: Mikhlin, Ye. B.; Korpusov, G. V.

ORG: None

TITLE: Extraction of rare earth elements of the cerium subgroup with the diisoamyl ester of methylphosphonic acid

SOURCE: Zhurnal neorganicheskoy khimii, v. 10, no. 12, 1965, 2787-2795

TOPIC TAGS: solvent extraction, rare earth element, phosphate ester, lanthanum, cerium, neodymium, samarium

ABSTRACT: The object of the work was a more systematic investigation of the extractive properties of the diisoamyl ester of methylphosphonic acid with respect to individual rare earth elements and of the possibility of its use for their separation. To avoid hydrolysis, extractions were made from weak nitrate solutions in which the content of free nitric acid did not exceed 0.2-0.25 N. The oxides of the rare earth elements (lanthanum, cerium, praseodymium, neodymium, samarium) contained more than 99.9% of the basic element. Also tested were binary mixtures and the following concentrates: Concentrate (A), rich in cerium earths, had the following composition (%): La-30; Ce-3.4; Pr-15.6; Nd-45.4; Sm-4; total yttrium elements-1.6. Concentrate (B),

Card 1/2

UDC: 541.183.3:546.654/.659-38

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ACC NR: AP6000763

rich in lanthanum, had the following composition (%): La-86.5; Ce-3.3; Pr-3.2; Nd-7.0. The acidity of the diisoamyl ester of phosphonic acid did not exceed 0.0005 N. The experiments were carried out at room temperature ($20 \pm 2^{\circ}\text{C}$) in separating funnels with a capacity of 150 ml. Distribution coefficients for the extractions were determined and are shown in tabular form. At an acidity of the aqueous phase of 0.1-0.2 N in HNO_3 , the extractive capacity of the diisoamyl ester of methylphosphonic acid attained 200-205 gram/liter of total rare earth oxides. Saturation of the organic phase for the diisoamyl ester of methylphosphonic acid is reached at lower equilibrium concentrations of rare earth elements in the aqueous phase than for tributyl phosphate. This makes it possible to use a lower concentration of the salting out agent- $\text{Al}(\text{NO}_3)_3$ --of the order of 2 or 3 N. Using 5 N $\text{Al}(\text{NO}_3)_3$ as a salting out agent, high values of the order of 2.5-3 are obtained for the separation coefficients for the vapors of lanthanum, cerium, neodymium, and samarium, which makes possible the use of the diisoamyl ester of methylphosphonic acid for the separation of rare earth elements of the cerium subgroup. Orig. art. has: 2 formulas, 5 figures, and 3 tables.

SUB CODE: 07/ SUBM DATE: 23May65/ ORIG REF: 009/ OTH REF: 002

jw
2/2
Card

MIKIMI, Dr. . .

"Patent-Non-Automatic Cell Changes in Experimental Contaminative Irradiation", Test, No. 2
Lavin et al., No. 1, Part. Mr. Lorraine Strobl, et al., Radiation Research
Physicians, -el. u.s.-,

MIKHLIN, Ye.G.

Infection of the larynx in brucellosis. Vest.otorinolaringol. 12 no.2:
78 Mr.-Ap '50. (CLM. 19:2)

1. Krasnoyarsk.

MIKHLIN, Y.E. G.
2297

Cases of damage to the organ of hearing in mumps Vestn, Otorino-laring. 1951, 1 (57-59)
A report on 7 cases: 2 with complete loss of cochlear and vestibular function, 2 with
bilateral and 3 with unilateral deafness. Daily lumbar punctures, intravenous injection
of 25% glucose solution, subcutaneous injection of pilocarpine (1: 1,000) and oral doses
of salicylates and hexamethylene tretamine (urotropine) were given in order to reduce the
intracranial and intralabyrinthine pressure. In 3 cases the auditory function was re-
stored but in 4 the treatment was without effect. The damage is ascribed to a menin-
geal serous labyrinthitis.

Salamun - Koper (XX, 6,7,8,11)

Pef. ... transmigrator

So: EXCERPTA MEDICA, Section VIII, Vol. 5, №.6, June 1952

MIKHLIN, Ye. G. Prof

PA 19686

Medicine - Selection of Flight Personnel Sep/Oct 51

"Selection of Professional Fliers (Correlation of Vestibulometric Data Taken in the Laboratory and in Flight)," Prof Ye. G. MIKHLIN, Chair of Diseases of the Ear, Nose, and Throat, Krasnoyarsk Med Inst

"West Oto-Rino-Laringol" No 5, pp 13-16

In tests used to select professional flight personnel too much attention need to be paid to the reactions of the semicircular canals in response to angular accelerations and not 19686

Medicine- Selection of Flight Personnel (Contd) Sep/Oct 51

enough to those of the otolithic app. Vojasik (1929) included both in his "OR" (otolithic reaction) tests given in Barani chair to study reactions. More attention should be paid to vegetative reactions (VR). Milov, who carried out animal expts involving detachment of otolithic membrane by centrifuging, recommends closer attention to objective, more advanced symptoms (pallor, cold sweat, nausea and later vomiting). Cumulative method is best for laboratory investigation.

19686

MIKHLIN, Ye.O., professor

~~Pathological changes in the hearing apparatus of white mice in experimental tuberculosis.~~ Vest.oto-rin. 17 no.1:59-60 Ja-F '55.
(MLRA 8:5)

1. Iz kliniki bolezney ukha, gorla i nosa Krasnoyarskogo meditsinskogo instituta.
(EAR--TUBERCULOSIS)

LUKOV, B.N., prof. (Kuybyshev); PETROV, V.I., dotsent (Moskva);
PAVLENKO, T.M., aspirant (Moskva); YERMOLAYEV, V.G., prof.
(Leningrad); ADO, A.D., prof.; VOVSI, M.S., prof.;
YERMOLAYEV, V.G., prof. (Leningrad); KUPRIYANOVA, N.A. (Kazan');
PETROV, G.I. (Moskva); DOLGOPOLOVA, A.V. (Moskva); SAYMAROV, P.P.,
prof.; BIKHOUISKIY, Z.Ye., prof.; MIN'KOVSKIY, prof. (Chelyabinsk);
KIMEL'CHONOK, I.P. (Irktusk); TEMKIN, Ya.S., prof. (Moskva);
MIN'KOVSKIY, A.Kh., prof. (Chelyabinsk); MIL'SHTEYN, T.N., doktor
med.nauk (Leningrad); TRUTNEV, V.K., zasluzhennyy deyatel' nauki,
prof.; TSYRESHKIN, B.D., kand.med.nauk (Moskva); SOBOL', I.M.,
prof. (Stavropol'); TUHIK, G.M. (Moskva); FRENKEL', M.M. (Moskva);
MAZO, I.L.; POKRYVALOVA, K.P.; PROSKURYAKOV, S.A., prof.;
ATKARSKAYA, A.A., prof.; GOL'DFARB, I.V., prof. (Izhevsk);
PORUBINOVSKAYA, N.M. (Moskva); RUDNEV, G.P., prof.; VOL'FSON, I.Z.,
prof. (Stalingrad); DOROSHENKO, I.T., prof. (Kalinin);
ROZENFEL'D, M.O., prof. (Leningrad); SHUL'GA, A.O., prof. (Orenburg);
~~MIKHLIN, Ya.G.~~, prof.; TRET'YAKOVA, Z.V. (Moskva); MANUYLOV, Ye.N.,
prof. (Moskva); DOROSHENKO, I.T., prof. (Kalinin); YERMOLAYEVA, V.G.,
prof.

Speeches in the discussion. Trudy gos. nauch.-issl. inst. ukha,
gorla i nosa no.11:79-87,129-146,179-186,233-248,311-333 '59.

(MIRA 15:6)

1. Chlen-korrespondent AMN SSSR (for Ado). 2. Direktor Moskov-
skogo gosudarstvennogo instituta ukha, gorla i nosa (for Trutnev).

(OTORHINOLARYNGOLOGY—CONGRESSES)

MIKHLIN, Ye.G., prof.

Indications for tympanoplasty in circumscribed labyrinthitis
with fistula. Vest.otorin. 22 no.5:49-52 8-0 '60.

(MIRA 13:11)

1. Iz kafedry bolezney ukha, gorla i nosa Krasnoyarskogo meditsinskogo instituta.

(LABYRINTH (EAR)—DISEASES)

MIKHLIN, Ye.G., prof. ; PAVLOV, Yu.V., ordinator.

Removal of impacted foreign bodies from the bronchi in children
using a drill. Vestn. otorinolaring. 25 no.3:102-103 '63
(MIRA 17:1)

1. Iz kafedry bolezney ukha, nosa i goria (zav. - prof. Ye.G.
Mikhlin) Krasnoyarskogo meditsinskogo instituta.

TRIFONOV, Dmitriy Konstantinovich; MIKHLINE, Ye.I., red.

[General economic laws] Osnovnye ekonomicheskie zakony
Leningrad, Izd-vo Leningr.univ., 1964.. 273 p.
(MIRA 18:2)

ZEMTSOVSKAYA, Tera Ivanovna; FUKHIM, Yel., red.

[Statistika i statisticheskaya metodika. Statistika prikladnaya i teoretičeskaya metoda posobie.
Leningrad, Izd-vo Leningr. univ., 1980. 53 p.
(MFA 17.8)]

"APPROVED FOR RELEASE: 06/14/2000

CIA-RDP86-00513R001134120016-2

A.C.S. MIKHLIN Ye. L.

Glase

Apparatus for making hollow glass products. E. L.
Mikhlin. Russ. 55,929. April 30, 1940. 22a. 12.
M.H.O.

APPROVED FOR RELEASE: 06/14/2000

CIA-RDP86-00513R001134120016-2"

BORISOGLEBSKIY, B.N. inzhener; MINKOV, V.P. inzhener, VEKSLER, G.M.
inzhener, TIKHILIN, Ye.L.; SALAMATOV, I.I. inzhener, redaktor;
STUPIN, A.E., redaktor; TIKHONOV, A.Ya., tekhnicheskiy redaktor

[Centrifuges; a catalog and reference book] TSentrifugi; katalog-
spravochnik. Moskva, Gos. nauchno-tekhn. izd-vo mashinostroit.
lit-ry, 1955. 90 p.
(MLRA 8:11)

1. Russiia(1923- U.S.S.R.)Ministerstvo mashinostroyeniya i priborostroyeniya.
(Centrifuges)

MIKHLIN, Ye.L., inzh.

Effect of the geometry of a rotor on the efficiency of a screw
settling centrifuge. Trudy NIIKHIMMASH no. 29:72-85 '59.
(MIRA 14:5)
(Centrifuges)

MIKHAIL VYATSKIY

EPP.
.392242

PYAT'DEZ' T. LIT. TRIT'YEGO SNEZHNOY POGORY. LENINGRAD, 1955.

41 P.

AT HEAD OF TITLE: VSESCYIVNOYE OCHKUSTV PO RAZ PODSTRELENIYU
"POLITICHESKIH I NAUCHNYKH IVANOV."

MIKHLINA, A. A.

MIKHLINA, A. A. "A New Disease in Central Asia, Virus of Cotton," in Results of the Work of the Station of Plant Protection of the All Union Order of Lenin Scientific-Research Institute of Cotton Production on the Study of Pests and Diseases of Cotton and Lucerne for 1939 (Auto-references and References.) Publishing House of the All Union Order of Lenin Scientific-Research Institute of Cotton Production, Tashkent, 1941, pp. 63-65, 464.04 T18

SO: Sira 31-90-53 15 Dec. 1953

VIZGACI, R.V.; VALTAN, A.F. (Vachanov, A.F.), KHLINA, I.A.

Infrared absorption spectra of esters of substituted benzene-sulfonic acids and phenol. Tr. fiz. khim. v. no.11:119-121, 1964.

CIA 174

1. Institut fiziki AN KarSSR, Kiev.

11(4), 7(3), 24(7)

AUTHORS: Mal'nev, A. F., Sklyar, V. T., Mikhлина, И. М., Puchkovskaya, G. A.,
Shulyak, L. I., Shevchenko, Ye. F.

SOV/48-23-10-8/39

TITLE: Investigation of the Composition of the High Molecular
Hydrocarbon Fractions of Petroleums of the Bitkovskoye Deposit
by Means of Infrared Absorption Spectra

PERIODICAL: Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, 1959, Vol 23,
Nr 10, pp 1192-1193 (USSR)

ABSTRACT: The present investigation was carried out in collaboration with
the laboratoriya geokhimii nefti Ukr. NIGRI (Laboratory for
Petroleum Chemistry of the Ukr. NIGRI). Investigations were
carried out of petroleums obtained from the boreholes 300, 310
and 350 of the Bitkovskoye deposit in the western Ukraine. First,
the solid fraction T was separated at 0 and -18°, and later the
aromatic fraction AT was separated according to the method of
Chernozhukov and Kazakova (Ref 1). The remainder of the solid
fraction OT was separated by carbamide complex formation
(complex-forming part KT, - non-complex-forming part NKT). The
remaining liquid fraction was chromatographically separated into
a paraffin naphthene fraction PNZh and a mono-, bi-, and poly-
cyclic aromatic hydrocarbon-containing fraction (1AZh, 2AZh and

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Investigation of the Composition of the High Molecular
Hydrocarbon Fractions of Petroleums of the Bitkovskoye Deposit by Means of
Infrared Absorption Spectra

nAZh). The fraction PNZh was further treated with carbamide and thiocarbamide and four components were obtained. The spectra were recorded in the range 2-15 μ by means of the vacuum infrared spectrometer of the type VIKS-3 (sample thickness 50 - 55 μ). In the following, a number of details concerning the spectra of the investigated fractions are given. The KT-spectra showed intense bands at 3.4 - 3.5, 6.82, 13.72 and 13.89 μ (corresponding to the oscillations of the CH₂-groups) and weak bands (CH₃) at 6.92 and 7.25 μ . The n-paraffins were characterized by the intense band at 13.89, the NKT-fraction by the 7.25 μ -band as well as that with 13.89 μ . The aromatic fractions had the following bands: 1AZh: 6.2, 122, 13.4, 13.8 and 14.3 μ (intense) and 9.6, 11.4 and 12.8 μ (weak). 2AZh: 6.2, 11.4, 12.2 and 13.4 as well as 12.8, 13.8 and 14.3 μ (weak). nAZh: 6.2, 11.4 and 13.4 as well as 9.6, 11.4 and 13.4 μ . The investigation results showed that the petroleums obtained from the various boreholes differ from one another. Thus, the T-fraction from the borehole 350

Card 2/3

Investigation of the Composition of the High Molecular SOV/48-23-10-8/39
Hydrocarbon Fractions of Petroleums of the Bitkovskoye Deposit by Means of
Infrared Absorption Spectra

contained more ramified paraffins than that from 310. The
petroleum of the former contained more aromatic, and that of the
latter more paraffin-hydrocarbons. There are 5 references,
3 of which are Soviet.

Card 3/3

MAL'NEV, A.F. [Mal'nev, A.F.]; MIKHLINA, I.M.

Using infrared absorption spectra for determining the quantity of oil in paraffins. Ukr.fiz.zhur. 6 no.6:859-861 N-D '61. (MIRA 16 5)

1. Institut fiziki i N UkrSSR, Kiyev.
(Spectrochemistry) (Paraffins)

VIZGERT, R.V. [Vishert, R.P.]; MAL'NEV, A.F. [Mal'niev, A.P.]; MIKHLINA,
I.M.

Effect of the nature and position of the substitute on the
infrared spectra of benzosulfochlorides and ethyl esters of
benzenesulfonic acid. Ukr.fiz.zhur. 7 no.5:512-514 My '62.
(MIRA 16:1)

1. Institut fiziki AN UkrSSR, Kiyev.
(Benzenesulfonic acid—Spectra)

VIZGERT, R.V.; MAL'NEV, A.F.; MIKHLINA, I.M.

Effect of the nature and position of substitutes on the infrared
spectra of benzenesulfonyl chlorides and benzenesulfonates. Izv.
AN SSSR. Ser. fiz. 27 no.7:969-973 '63. (MIRA 16:8)
(Benzenesulfonyl chloride--Spectra)
(Benzenesulfonic acid--Spectra)

OBOLONCHIK, V.A.; MIKHLINA, T.M.

Cerium selenides and oxyselenides. Zhur. prikl. khim. 38 no.7:1451-1456
J1 '65. (MIRA 18:7)

"APPROVED FOR RELEASE: 06/14/2000

CIA-RDP86-00513R001134120016-2

APPROVED FOR RELEASE: 06/14/2000

CIA-RDP86-00513R001134120016-2"

MIKHLINA, S.YE.

Determination of dianhydrophthalimide in rubber mixtures.
S. R. Mikhlin, Sov. Pat. No. 347117; Russ. Korr. Tsvetnoy Ind.,
1954, No. 6, 70-3; Referat Zbir., Khim. 1955,
Abstr. No. 50718.—Diphenylquazidine (DPO) was detd. by
neutralization with excess 0.1N HCl and subsequent titration
with 0.1N alkali soln. in the presence of methyl orange.
The method was verified with DPO samples and with
mixts. of synthetic rubber and DPO. In the latter case,
the precision is within 5.00%. The solv. of DPO and
mercaptobenzothiophene was also studied. Best solvents are
Me and Et alcohols. N. Yashin //

4
482c (4)
2 May

L 12020-65 EWT(m)/EMP(b) RDM/JD

ACCESSION NR: AP4047995

8/0073/64/030/010/1037/1039

AUTHOR: Obolonchik, V. A.; Mikhлина, Т. М.

TITLE: Preparation and properties of rhenium selenide 8

SOURCE: Ukrainskiy khimicheskiy zhurnal, v. 30, no. 10, 1964, 1037-1039

TOPIC TAGS: rhenium selenide, rhenium diselenide, rhenium diselenide preparation, rhenium diselenide property, hydrogenation catalyst, hydrogen selenide synthesis

ABSTRACT: The preparation and chemical properties of rhenium diselenide, ReSe₂, have been studied to supplement available data on rhenium selenides, which are considered good catalysts of hydrogenation. Two methods of preparing pure ReSe₂ were developed using either the reaction of rhenium metal with hydrogen selenide or the reaction of ammonium perrhenate with hydrogen selenide. The synthesis of pure hydrogen selenide from elements and the preparation of spectroscopic-grade ammonium perrhenate from rhenium metal were described. The same

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J. 12020-65

ACCESSION NR: AP4047995

laboratory apparatus was employed for the study of both solid-state reactions, which produced ReSe_2 . The hydrogen-selenide synthesis and the preparation of ReSe_2 were carried out in two sections of the same reaction tube, heated to 500 and 700°C, respectively. Experimental data indicated that both methods of preparation required the same temperature but different optimum time. Chemical and x-ray analysis were used to determine the product composition. A method was developed to determine rhenium in the product by hydrogen reduction of ReSe_2 at 900°C. The chemical properties of pure ReSe_2 were determined. The ReSe_2 products were found to be active catalysts of the hydrogenation of n-butane and butenes, which produces divinyl. Orig. art. has: 1 figure, 3 tables and 2 formulas.

ASSOCIATION: Institut problem materialovedeniya AN UkrSSR (Institute for the Study of Materials, AN UkrSSR)

SUBMITTED: 09 Sep 63 ATD PRESS: 3122 ENCL: 00

SUB CODE: CC, IO NO REF. SOV: 002 OTHER: 001

Card 2/2

L 57869-65 EWT(1)/EMT(m)/ENG(m)/T/EWP(t)/EWP(b)/EMA(h) P2-6/Peb IJP(c)
RDW/JD/JG/AT

ACCESSION NR: AP5017774

UR/0080/65/038/007/1451/1456
566.655'23

30
B

AUTHOR: Obolonchik, V. A.; Mikhлина, Т. М.

TITLE: Cerium selenides and oxysele-nide

SOURCE: Zhurnal prikladnoy khimii, v. 38, no. 7, 1965, 1451-1456

TOPIC TAGS: rare earth selenide, cerium selenide, cerium monoselenide, cerium sesquiselenide, cerium oxysele-nide, selenide preparation, semiconductor material

ABSTRACT: Preparation of cerium selenides has been studied in the first of a series of studies on the selenides of rare-earth elements. The study was initiated because of the possible use of these selenides in semiconductor and other new technological fields and because very few data are available in the literature. Direct synthesis from the elements was discarded as unpractical, partly because of the scarcity of pure rare-earth elements. Two indirect methods were explored. One, the reaction of pure cerium dioxide with pure selenium in solid phase produced at 1100—1200°C and under specified conditions nearly pure cerium oxysele-nide Ce₂O₂Se. The data obtained at temperatures to 1350°C suggested that at even higher temperatures preparation of cerium selenide CeSe would be possible. Electric resistivity of the compacted and slightly sintered Ce₂O₂Se sample was found to be of the order of

Card 1/2

L 57869-65

ACCESSION NR: AP5017774

4×10^5 ohm·cm. The second method consisted of reacting pure cerium dioxide with hydrogen selenide which was synthesized in the reactor. Nonstoichiometric cerium monoselenide was obtained by this reaction at 1100°C after 2 hr. At longer (5–6 hr) reaction time and 1100–1200°C, a stoichiometric cerium selenide Ce_2Se_3 was obtained. The latter can be converted into CeS by heating in hydrogen stream or in vacuum.
Orig. art. has: 3 figures and 3 tables.

{JK}

ASSOCIATION: none

SUBMITTED: 10Jun62

NO REF Sov: 000

ENCL: 00

SUB CODE: IC,GC

OTHER: 006

ATD PRESS: 4038

Card 2/2

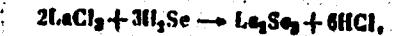
L III48-66 EWT(m)/ETC(F)/EMG(m)/EMP(t)/EMD(b) IJP(c) RDW/JD

ACC NR: AP6000688

SOURCE CODE: UR/0080/65/038/009/2100/2103

AUTHOR: Obolonschik, V. A.; Mikhline, T. M.ORG: Institute for the Problems of Materials of the AN UkrSSR
(Institut problem materialovedeniya AN UkrSSR)TITLE: Lanthanum selenides and oxyselenidesSOURCE: Zhurnal prikladnoy khimii, v. 38, no. 9, 1965, 2100-2103TOPIC TAGS: selenide, selenium compound, lanthanum compound, chemical reaction, hydrogenation

ABSTRACT: A series of experiments was made on the reaction of lanthanum chloride and hydrogen selenide. This method can be illustrated by the following equations:



Hydrogen selenide was synthesized from 99.998% pure selenium in a quartz reactor and was passed over lanthanum powder at temperatures on the order of 300-1200°C. Results of the reaction are shown in tabular form.

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UDC: 546.654'23

L 11148-66

ACC NR: AP6000688

To obtain lanthanum diselenide, the temperature was gradually increased to 750-800°C and was held at this level for 2 hours. To obtain hydrogen sesquiselenide, the temperature was raised to 1100-1150°C and held at this level for 2-2.5 hours. To study the stability of La₂Se₃ and La₂Se₄, and to obtain a phase with a lower content of selenium, studies were carried out on the hydrogenation of lanthanum sesquiselenide. At temperatures of 350-380°C, there was obtained a black product corresponding by analysis to lanthanum sesquiselenide. This phase is stable up to a temperature of 550°C; at higher temperatures (on the order of 900°C) there is evolved the known brick-colored lanthanum sesquiselenide. The data indicate that lanthanum oxyselenide, La₂O₂Se, is obtained in the reaction of lanthanum oxide with selenium in an inert atmosphere at temperatures of 1100-1200°C. The action of hydrogen selenide on lanthanum oxide or chloride at 700-800°C leads to the production of La₂Se₄, while at temperatures of 1100-1200°C it leads to the production of lanthanum sesquiselenide, La₂Se₃. On heating lanthanum sesquiselenide to 1200°C in a stream of hydrogen, the selenium content decreases to a limiting value corresponding to lanthanum sesquiselenide. Orig. art. has: 3 figures and 2 tables.

SUB CODE: 07/ SUBM DATE: 20Jul63/ ORIG REF: 001/ OTH REF: 003

AC
Card 2/2

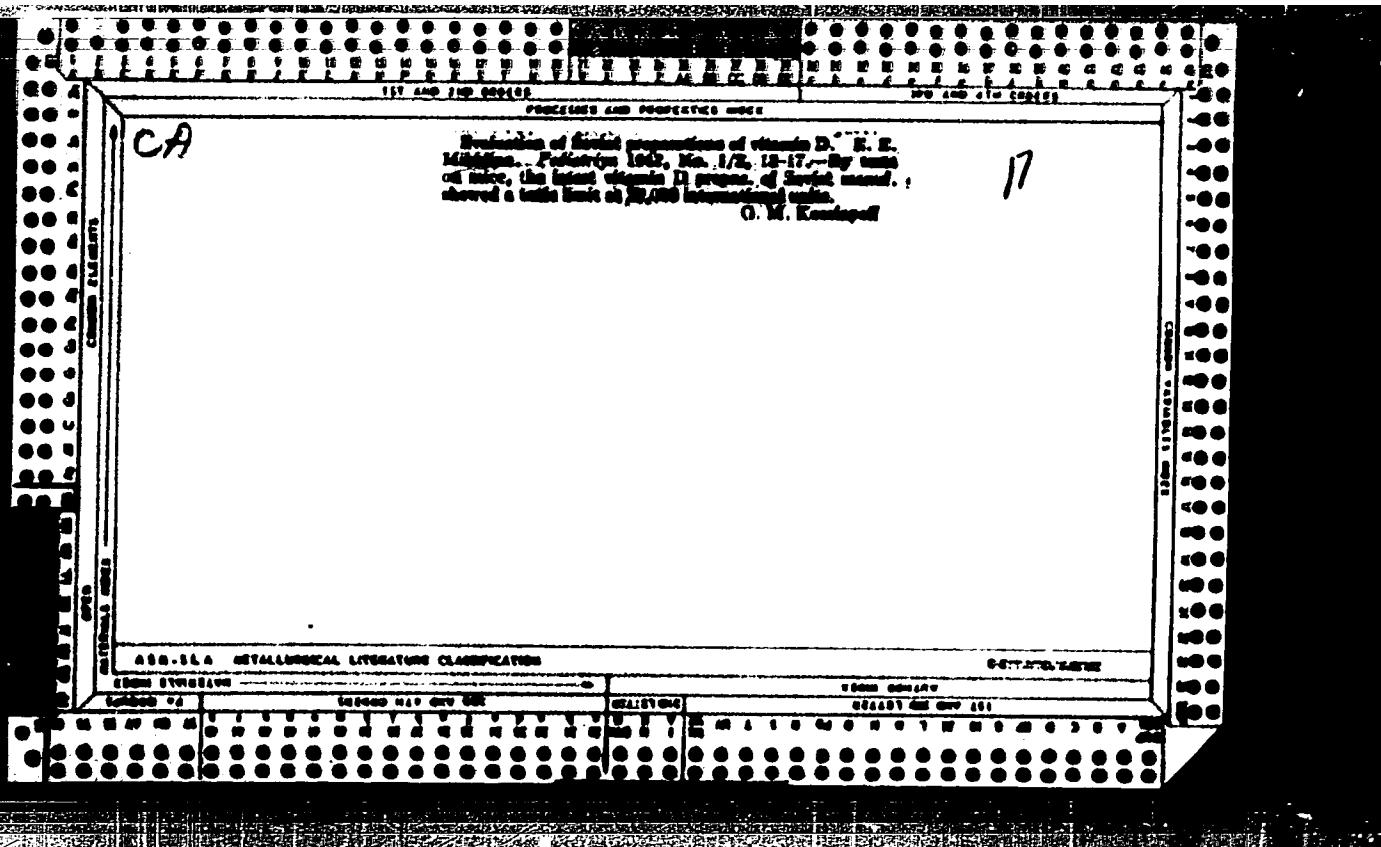
KOLCHINSKAYA, L.M.; RUSOVA, L.A.; MIKHLINA, V.V.

Introducing the manufacture of No.10, 7 nylon cord. Khim. volok.
no.1;7-8 '62. (MIRA 18:4)

LENIN, M.P., and many others like him, think that the present
country is terrible; ~~THEIR~~ ~~IDEAS~~ are very bad, they say.
They, too, say communists

deserve the strongest demands. They are the most important
part of the country, because they are the best people in the
country.

• Every day there are more communists in the country, and
they are growing rapidly.

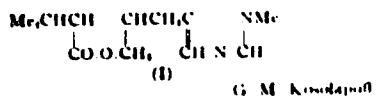


Promergane alkaloids. XX. Synthesis of racemic isopropylparaconic acid. A. G. Naradov and B. N. Mikhlin (Lomonosov Inst. Fine Chem. Technol., Moscow), *J. Russ. Chem. Soc. (U.S.S.R.)* 17, 1718-27 (1947) (in Russian); *C. A.* 40, 31477. - Isopropyl acid (100 g.) and PCl_5 were heated to 100° , then treated with 300 g. H_2O_2 and heated until decolorized; the soln., slowly added with stirring to 147 g. abs. EtOH and treated with ice, gave 70.7% *Et*- α -bromoconiacate, bp $91-93^\circ$. This (146 g.), added to the reaction mixt. of 100 g. di- Et malonate, 15.2 g. Na, and 170 g. $\text{EtO}(\text{H})$, then heated to 100° 12 hrs., freed of $\text{EtO}(\text{H})$, and washed with H_2O , gave 68% $\text{MeC}_2\text{Cl}(\text{CH}_2\text{CO}_2\text{Et})(\text{CH}_2\text{CO}_2\text{Et})$, bp $172-3^\circ$. This (83 g.), 165 cc. concd. HCl, and 151 cc. water, refluxed until hydrolysis was complete, gave an colourless *disopropylconic acid*, m.p. 107° . This (27 g.) and 54 g. Ac_2O heated 4 hrs., gave 86% *isopropylconic anhydride*, bp 100° , a solid, which (21 g. in 100 cc. EtOH) was sld with HCl , heated on a steam bath 2 hrs., then cooled to room temp., and treated with 100 cc. NaHCO_3 , gave 70% *di-Et isopropylsuccinate*, bp 129° . Thus (24.4 g.) and 0.3 g. HCO_2Et were added to the EtONa from 2.0 g. Na and 6.2 g. $\text{EtO}(\text{H})$ in 78 cc. dry Et_2O , allowed to stand 3 days in the cold and 2 days at room temp., then treated with ice water, and the aq. soln. was sepd. and ext'd. with Et_2O ; the aq. soln. was carefully acidified to Congo red with 10% H_2SO_4 , ext'd. with Et_2O , and the ext'ns. poured onto amalgamated Al, allowed to react 3-4 hrs., and filtered, and the combined filtrate and Et_2O washings were dried, to give 72% mixed stereoisomers. Et_2O washings, bp $100-7^\circ$ and $175-9^\circ$; the 1st is the stable form, a.i. 1.445, $d_2^{20} 1.1058$; the 2nd crystallizes on cooling and m.p. $45-7^\circ$. The stable form (21.7 g.), boiled 16 hrs. with 4 parts of 1.1 HCl and cooled *in vacuo*, gave 81.0% iso-propylparaconic acid, in $\text{K}_2\text{Cr}_2\text{O}_7$ (from water or benzene). This (8.1) and 20 g. NaCl , heated 5 hrs. to $101-9^\circ$ gave 51% 90.2% acid chloride, bp $109-11^\circ$. This (3.3 g.), slowly added with stirring at -11° to the CH_2N_2 from 21 cc. H_2O , stirred 3 hrs. at -10° , and concd. *in vacuo*, gave 50.2% *disomethyl isopropylparaconic acid*.

isopropyl ketone, Me₂C(CH₂CO₂Et)₂

127.8-8.8°. This (10.8 g.) in 175 cc. atm. $\text{EtO}(\text{H})$ was added to 0.7 g. fresh $\text{Ag}(I)$ in 50 cc. atm. KOH with stirring, then kept 2 hrs. at 50° and refluxed 1 hr., after filtration, the soln. was diluted to give 50.5% *Et isopropylparaconate*, b.p. 101.3° , n_D²⁰ 1.467, d₂₀²⁰ 1.062. This (4.6 g.) and 8 parts 30% HCl, refluxed 10 hrs., gave 50.5% *isopropylparaconic acid*, m.p. 110° . This with SOCl_2 gave 64.5% chloride, bp $102-3^\circ$, which was converted, as described above, into *disomethyl isopropylparaconyl ketone* (80.3%), m.p. $84-3^\circ$. This (from 3.0 g. acid), 25 g. $\text{Ag}(II)$, and 6 g. $\text{Ag}(I)$, heated 8 hrs. to 100° , then boiled 2 hrs., gave *disomethyl isopropylparaconyl ketone* (81.4%), bp $181-3^\circ$, d₂₀²⁰ 1.104, n_D²⁰ 1.411. This (0.92 g.) in 20 cc. Hg was added to 2.7 g. Cu acetate and 3.5 g. 40% formalin in 75 cc 25% NH_4OH and the melt heated to boiling over 1 hr. treated 0.5 hr. with H_2S , filtered, brown with air, treated with 1/2 vol. concd. HCl, and boiled 0.5 hr., after evapn to dryness, the residue was taken up in a little HgCl_2 , treated with NH_4OH , and ext'd. with CHCl_3 , evapn of the ext'd. gave crude *isopropylparaconide*; this, neutralized with 1:16 HNO_3 to weak Congo red color and concd. *in vacuo*, gave *isopropylparaconic nitrate*, m.p. $124.5-5.5^\circ$ (from atm. EtOH); 40% K_2CO_3 gave 88.8% of the free base, m.p. $88-9^\circ$ (from EtOH); HCl add (73%) yield by evapn. The aq. HCl soln. of the base, very hygroscopic. The HCl salt (0.3 g.), 0.24 g. 35% formalin, and 0.2 cc. H_2O , heated until clear, then treated with 0.12 g. HCO_2Et , heated 15 hrs. to 101° , evapd. *in vacuo*, and the residue neutralized by 1:16 HNO_3 , followed by evapn. to *isopropylparaconic nitrate* (20.3%), m.p. $134-9^\circ$; the free base was isolated as a yellow oil. The above methylation method was also checked as follows: 1.5 g. 1-(hydroxymethyl)imidazole-HCl, 2 g. 35% formalin, in 1.5 cc. water were heated until clear, treated with 1 g. HCO_2Et , and heated 15 hrs. on a 100 rpm bath; after evapn. *in vacuo* and treatment of the residue with EtOH , the residue of 1-(hydroxymethyl) imidazole-HCl was recovered.

methyl iodide was converted to the picrate (m. 3° yield).
The picrate was converted to the HCl salt, m. 92-4°, which
on mixing with the HCl salt of the starting material gave
a m. p. of 93-7°.



MIKHLINA, YE. YE.

USSR/Chemistry - Antitubercular Drugs

Jan 52

"Synthesis of Certain Aminoguanidine Derivatives,"
M. N. Shchukina, Ye. Ye. Mikhlina, All-Union Sci
Res Chem-Phar Inst imeni S. Ordzhonokidze

"Zhur Obshch Khim" Vol XXII, No 1, pp 132-135

Prepd a number of substituted benzalguanyls and
their N⁴-alkyl derivs. Found from tests at Div of
Chem Therapy, All-Union Sci Res Chem-Phar Inst that
these compds possess tuberculostatic activity but
are toxic.

20725

MIRKINA, E. E.

Chemical Abst.
Vol. 48
Apr. 10, 1954
Organic Chemistry

Synthesis of 2,3-substituted quinuclidines. I. V.
Rimant and Yu. P. Mikhlin. S. O. Gromova. All-USSR
Chem. Pharm. Inst., Moscow. Zhar. Obshch. Khim. 23,
N21-24 (1953). To 1.27 g. Na in 20 ml. alc. Et₂O was
added 8.7 g. CH₃CO₂Et₂ and 0.0 g. Et 4-pyridine acetate,
and the mixt. stirred 1 hr. at 0° and treated with very dil.
AcOH and extd. with Et₂O, yielding 94.2% Et β -[bis(ethoxycarbonyl)methyl]- δ -pyridinepropionate, b.p. 173-5° (cone. decomp.); HCl salt (I), m. 121-2° (from EtOH-Et₂O). This
(4.4 g.) refluxed 8 hrs. with 44 ml. concd. HCl gave 7-(δ -
pyridyl)glutaric acid-HCl, which, heated 3 hrs. with 35 ml.
5% alc. HCl, concd. *in vacuo*, treated with K₂CO₃, and extd.
with Et₂O, yielded 84.4% di-Et ester, b.p. 140-8°. I reduced
over Pt oxide in EtOH to Et β -[bis(ethoxycarbonyl)methyl]- δ -
piperidinepropionate, isolated as the HCl salt (II), tarry-
like mass; the free base is amorphous; treatment with Ac₂O
gave the 1-Ac deriv., b.p. 200-7°. II (from 47.8 g. pyridine
analog) in dry CHCl₃, treated over 8 hrs. with 20.4 g. Br in
CHCl₃, allowed to stand 12-14 hrs., concd., treated with
H₂O and K₂CO₃, and extd. with Et₂O yielded crude Et
 β -[bis(ethoxycarbonyl)bromomethyl]- δ -piperidinepropionate,
which, boiled 2 hrs. with 300 ml. pyridine, concd., and
treated with 50% K₂CO₃ gave 72% Et 2,2-dicarbethoxy-3-
quinuclidineacetate, b.p. 147-8°; methiodide, m. 139-41°
(from EtOH-Et₂O). The ester refluxed 10 hrs. with concd.
HCl gave 2-hydroxy-3-quinuclidineacetic acid-HCl (III),
decomp. 254-5° (from eq. Me₂CO); the pure product de-
comp. 254-5° (from EtOH-Et₂O). This (0.4 g.) 1.1 g.
Ag₂O and 6 ml. H₂O shaken 2 hrs.; dried, heated to the b.p.,
filtered, satd. with H₂S, filtered, and evapd. gave 43.8% 2-
carboxy-3-quinuclidineacetic acid, m. 205-7°; an 87.0%
yield is obtained with 5% alc. NH₃. III (7 g.) heated with 100
ml. SOCl₂ 10 hrs. at 70°, freed of SOCl₂, and the resulting
acid chloride-HCl (IV) refluxed 3 hrs. with EtOH gave
82.7% di-Et 2-carboxy-3-quinuclidineacetate, b.p. 128°, n_D
1.4797, (also obtained from III and 5% alc. HCl refluxed 6
hrs.); methiodide, m. 140-1° (from EtOH-Et₂O). The
ester (8.2 g.) in Et₂O treated with 4.64 g. LiAlH₄ suspended
in Et₂O, boiled 1 hr., and treated with 8 ml. H₂O gave 88.7%
2-hydroxymethyl-3-(2-hydroxyethyl)quinuclidine, b.p. 150-7°;
HCl salt, hygroscopic solid. The latter (5.22 g.) in dry

2/2

Rubtsov, M. V.
CHCl₃ treated with 18 ml. SOCl₂, boiled 0.8 hr., and concd.
in vacuo, gave 84% 3-chloromethyl-3-(2-chloroethyl)quinuclidine-HCl, m. 139-40°; with 50% K₂CO₃ it gave the free
base, b.p. 120-2°, which forms a methiodide, m. 138°. IV
(from 3 g. acid HCl salt) and 40 ml. Et₃NCH₂CH₂OH kept
3.5 hrs. at 80-8° gave 63% bis(3-diethylaminoethyl) ester,
b.p. 187-9°, of 3-carboxy-3-quinuclidineacetic acid; *isomeric*
iodide, decomp. 197-8°. G. M. Kosolapoff

(3) 4

L 30386-66 EWT(m)/EWF(.)/ETI LIP(.) JDP(G)
ACC NR: AP6019663

SOURCE CODE: UR/0073/66/032/006/0567/0572

AUTHOR: Obolonchik, V. A.; Mikhлина, Т. М.

ORG: Institute of the Science of Materials, AN UkrSSR (Institut problem
materialovedeniya AN UkrSSR)

TITLE: Selenides of praseodymium, neodymium, samarium, and europium

SOURCE: Ukrainskiy khimicheskiy zhurnal, v. 32, no. 6, 1966, 567-572

TOPIC TAGS: selenide, ~~polyselenide~~, praseodymium, neodymium, samarium, europium,
~~CHEMICAL REACTION~~

ABSTRACT: Polyselenides of praseodymium, neodymium and samarium of the Me_2S_4 type were prepared by reacting the respective metal oxides with hydrogen selenide at 750—800°C. Sesquiselenides of these metals were prepared by similar reactions at 1100—1200°C, or by thermal decomposition of the polyselenides in hydrogen or in an inert gas. The reaction of europium with hydrogen selenide at 1100°C yielded europium monoselenide ($EuSe$); europium selenides with a higher selenium content were not obtained. The preparative procedure of the selenides, their properties, their stability in various media, and the preparation of compacted selenide specimens are described in the source. Orig. art. has: 3 figures and 3 tables. [B0]

SUB CODE: 07/ SUBM DATE: 03Dec64/ OTH REF: 005/ ATD PRESS: 5017

Card 1/1 CC

UDC: 546.655.23

MIKHINA, Ye. Ye.

USSR/Chemistry - Pharmaceuticals,
Bactericides

Jul 53

"Synthesis of Substituted Hydroxy- and Dihydroxydiphenylmethanes," M. V. Rubtsov, Ye. Ye. Mikhlina and L. A. Pelenitsina, All-Union Sci-Res Chem-Pharm Inst im Ordzhonikidze

Zhur Obshch Khim, Vol 23, No 7, pp 1209-1214

Simplified the methods found in the literature for the prepn of 4,4'- and 2,4'-dioxydiphenylmethanes. Synthesized a series of derivs of hydroxy- and dihydroxydiphenylmethanes and tested their chemotherapeutic activity.

272720

Mikhailina, Ye. Ye.

USSR/Chemistry - Drugs

Sep 53

"Aminoalkyl Derivatives of Quinuclidine," M. V. Rubtsov, Ye. S. Mikitskaya, Ye. Ye. Mikhailina, A. D. Yanina, and V. Ya. Furshtatova, All-Union Sci-Research Chemico-Pharmaceut Inst im Ordzhonikidze

Zhur Obshch Khim, Vol 23, No 9, pp 1555-1559

A number of substituted 2-aminomethyl quinuclidines and 2-aminomethyl-3-(⁻-aminoethyl)-quin-

268T33

MIKHLINA, Ye Ye

260T6

USSR/Chemistry - Pharmaceuticals

21 Feb 53

"Synthesis of 2,3-Disubstituted Quinuclidines," M. V. Rubtsov and Ye. Ye. Mikhlina, All-Union Sci-Res Chem-Pharm Inst im S. Ordzhonikidze

DAN SSSR, Vol 88, No 6, pp 1003-1006

Synthesized β -carboxyquinuclidyl-(3)-acetic acid, some of its derivs, and conversion products starting with the ethyl ester of β -dicarboxymethyl- β -[pyridyl-(4)]-propionic acid. Presented by Acad V. M. Rodionov 27 Nov 52.

260T6

Mikhailov, Ye. Ye.

V Synthesis of stereoisomeric 2-phenyl-3-(4-piperidyl)butanoic acids
 Dimitriev, M. V., Rubtsov, V. P., Mikhailova, and V. V. A.
 Birshteinova (S. Ordzhonikidze Institute of Chemical Research)
 Chem.-Pharm. Inst., Moscow), Zavod. Obshchel. Khim. 24,
 2059-02-1954; cf. C.A. 48: 3073a.—A mixt. of EtONa
 (from 2.8 g. Na and 40 ml. EtOH), 18.50 g. $\text{PbCl}_2\text{CO}_2\text{Et}_2$,
 and 20 g. Et-3-(4-piperidyl)acrylate heated 2 hrs. at 60°,
 and treated with very dil. AcOEt gave 90% di-Et 2-phenyl-3-
 (4-piperidyl)glutarate, m. 100° (from petr. ether), hydrolyzed
 with boiling 1:1 HCl 4 hrs. gave the free acid, decomp. 231-
 2°, insol. in org. solvents. Hydrogenation of the di-Et ester
 over a Pt oxide catalyst in PtO_2 contg. dry HCl required
 140 hrs., yielding the corresponding 3-(4-piperidyl) analog
 (I), m. 70-80°, as a monohydrate; dilute results in loss of
 H_2O , yielding the anhyd. free ester, b.p. 172°, which readily
 picks up 1 H_2O . The product heated with Ac_2O 1 hr. gave
 di-Et 2-phenyl-3-(1-acetyl-4-piperidyl)glutarate, b.p. 215-18°,
 m. 85°. Hydrolysis of I required 16 hrs. In concd. HCl gave
 a less-sol. isomer of the free acid HCl salt (II), decomp. 230-
 1°, and a more-sol. isomer (III), decomp. 178° (from $\text{EtOH}-$
 Et_2O). The former isomer with alc. NH_2 gave a free acid
 decomp. 210-12°, while the latter gave a free acid isomer
 decomp. 272°. Heating II *in vacuo* 16 min. at 230° con-
 verts it to III, isolated as the HCl salt. The di-Et ester
 with Br in CHCl_3 at 20° gave di-Et 2-phenyl-3-(4-piperidyl)-
 glutarate bromide HBr salt, $\text{CaH}_2\text{O}_2\text{NBr}$, decomp. 145°.

O. M. Kosolapoff

7/14/1961 Rina YEX.

62 Synthesis of substituted derivatives of 2-aminobenzoic acids.
M. V. Rukutov, B. Z. Miller, and V. Ya. Pashkovna (S. Ordzhonikidze Institute of Chemical Physics, USSR Academy of Sciences, Kosygin Str. 4, Moscow, 117334, U.S.S.R.)
Received January 29, 1964; revised, May 24, 1964; cf. C.A., 61, 11166.
The synthesis of 2-aminobenzoic acids (I) was carried out by reduction of 2-carboxybenzoic acid (II) in H_2O at room temp. gave 11.2% of crude 2-carboxybenzoic acid, 3-carboxylic acid (I)
(from 100% $\text{BROH} \cdot \text{RuO}_4$) and some 2-carboxy- α -quinones (from 40% $\text{BROH} \cdot \text{RuO}_4$).
When the CHCl_3 solution of 2-carboxybenzoic acid (II) was treated with 47% SO_3 in CHCl_3 (2 hr), followed by treatment with 47% NH_3 in BzO solution (50.0% 2-carboxybenzoic acid (II)), m. 103-8° (from RuO_4), it was proposed the structure of 2,6-diaminobenzoic acid (III).
This was confirmed by its conversion to 2,6-diaminobenzoic acid (IV), m. 153-6°, and the disappearance of absorption bands at 1650, 1600, and 1500 cm⁻¹.
The reduction of 2,6-diaminobenzoic acid (IV) with 10% NaBH_4 in CHCl_3 gave 70.2% of 2,6-diaminobenzoic acid (V) (m. 125-127°), m. 128-8°, and 50.5% of 2,6-diaminobenzoic acid (VI) (m. 100-2°, m. 105-7°, m. 111 (3:1)).
Treatment of 2,6-diaminobenzoic acid (V) with 24% NaBH_4 in $\text{BzO-C}_6\text{H}_6$ gave 78.7% of 2,6-diaminobenzoic acid (VI) (m. 100-2°, m. 105-7°, m. 111 (3:1)).
Similar reduction of the older product (VI) gave 65.6% of 2,6-diaminobenzoic acid (V) (m. 125-127°, m. 128-8°, m. 100-2°, m. 105-7°, m. 111 (3:1)).
It was shown that the reduction of 2,6-diaminobenzoic acid (V) with 10% NaBH_4 in CHCl_3 gave 80.4% of 2,6-diaminobenzoic acid (VI) (m. 100-2°, m. 105-7°, m. 111 (3:1)).
Thus, the reduction of 2,6-diaminobenzoic acid (V) with 10% NaBH_4 in CHCl_3 gives 65.6% of 2,6-diaminobenzoic acid (VI) (m. 100-2°, m. 105-7°, m. 111 (3:1)).
Treatment of 2,6-diaminobenzoic acid (VI) with 24% NaBH_4 in $\text{BzO-C}_6\text{H}_6$ gave 78.7% of 2,6-diaminobenzoic acid (V) (m. 125-127°, m. 128-8°, m. 100-2°, m. 105-7°, m. 111 (3:1)).
Thus, the reduction of 2,6-diaminobenzoic acid (VI) with 24% NaBH_4 in $\text{BzO-C}_6\text{H}_6$ gives 65.6% of 2,6-diaminobenzoic acid (V) (m. 125-127°, m. 128-8°, m. 100-2°, m. 105-7°, m. 111 (3:1)).

②

Q.M. Kecce

M. Khlin, YE/E

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✓ New paths of synthesis of 2-quinoxalidinocarboxylic acid.
 M. V. Rubtsov and E. N. Mikulin (S. I. Vavilov All. Union Chem. Pharmacol. Inst., Moscow). Zhur. Obshch. Khim. 25, 2303-10 (1955).—Refluxing 16 g. 4-(2-hydroxy-3,3-dichloropropyl)pyridine (I) with 40 ml. EtOH, 10 ml. H₂O, and 19 ml. AcOH and 9 g. Zn dust added over 20 min., gave 0.07 g. 4-(1-hydroxy-3,3-dichloropropyl)pyridine, m. 134-0°. Hydrogenation of I over Pt in alc. HCl gave 91.8% corresponding piperidine-II (II). HCl, decomp. 191-6°; II, decomp. 150-60° (from EtOH); HBr salt, decomp. 189-0°. II heated with MeI in EtOH gave 51.8% N-Me deriv., m. 172.5-3.5° (from EtOH). Similarly, PhCH₂Cl gave 71.8% N-benzyl deriv., m. 150-1°. Heating II with POCl₃-PCl₃ at reflux 2 hrs., wash. in Et₂O and washing the ext. with H₂O and NaHCO₃, gave II phosphate, C₁₂H₁₄O₄N₂, m. 210°. II heated with 68% HBr in sealed tube 3 hrs. at 120° gave 4-(2-bromo-3,3-dichloropropyl)piperidine-II Br, decomp. 160-3.5°, in 40.6% yield. This (1 g.) in 12 ml. H₂O, treated with 15 ml. 15% Na₂CO₃ and 20 ml. CHCl₃ and shaken 3 hrs., gave a little 3-chloromethylquinoxalidine, m. 141° (m.p. H₂O, decomp. 230° (from EtOH-HCl)). II in EtOH heated with alc. KOH 4 hrs. at 40°, treated with CO₂, filtered, evapd. and taken up in 10% HCl gave 61.8% β -4-piperidylbutanoic acid-HCl·H₂O, decomp. 202-4° (from EtOH-Et₂O); its ester, decomp. 230° (from EtOH-Et₂O); free ester base (III), b.p. 147-8°, n_D²⁰ 1.4628, on standing changes to a solid mass, m. 230-1°, indicating that this substance is a polymer which can be depolymerized by long heating with alc. HCl. III with BaCl₂

(OVER)

NEW PATHS OF SYNTHESIS . . .

In pyridine gave 85.7% the *N*-Bs derivative, b.p. 178-79°, n_D 1.518. III evapd. with 20% HCl-EtOH, then treated with BzCl , at 63°, evapd., treated with aq. K_2CO_3 , followed by pyridine, gave a low yield of an ester, b.p. 137-47°, which refluxed with 17% HCl gave a poor yield of 2-quinuclidine-carboxylic acid, decimp. 226°, while the residue yielded II. II in pyridine treated with BzCl , then treated with 10% HCl, gave 93.8% *N*-Bs derivative, m. 177-9°. (from EtOH). This treated with alc. SOCl 48 hrs. gave $\text{K} \beta-(N\text{-benzoyl-4-piperidyl})$ lactate, which treated with aq. AgNO_3 gave the Ag salt, a solid which darkens in light. This with Et_2O refluxed 0 hrs. gave 87.6% $\text{Et} \beta-(N\text{-benzoyl-4-piperidyl})$ lactate, b.p. 173°, n_D 1.518, which refluxed with 17% HCl 4 hrs. gave after the usual treatment 63.7% $\text{Et} \beta-(4\text{-piperidyl})$ lactate, b.p. 147-5°, n_D 1.4028, which polymerizes on standing; 77.3% is obtained if the refluxing is performed 2 hrs. with 20% EtOH-HCl.

G. M. Knutson

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PM 4

Mikhлина, Е. Е.

USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61542

Author: Rubtsov, M. V., Mikhлина, Е. Е.

Institution: None

Title: Preparation and Properties of Ethyl-N-(quinuclidyl-2)-urethane

Original

Periodical: Zh. obshch. khimii, 1956, 26, No 1, 135-138

Abstract: Described is an attempt to synthesize 2-aminoquinuclidine from quinuclidine carboxylic acid-2 (I) over its hydrozide (II) according to Curtius. On interaction of II with iso-C₅H₁₁-ONO (III) in alcoholic solution of HCl there is obtained ethyl-(IV) and in the solution of HCl in iso-C₅H₁₁OH respectively the isoamyl (V)-N-(quinuclidyl-2)-urethane; together with IV as a by-product were isolated ethyl-(VI) and with V the isoamyl (VII) ester of I. On saponification of IV by action of dilute HCl was obtained NH₄Cl and amorphous polymer of dehydroquinuclidine (VIII). On heating of IV with C₆H₅CH₂NH₂ (IX) to 100° evidently takes place

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USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61542

Abstract: transacylation as a result of which is formed benzyl-urethane (X) and 2-aminoquinuclidine; the latter split off NH₃ and is converted to VIII. On heating of IV with IX (or with C₆H₅NH₂) to 180° NH₃ is evolved and together with VIII there is formed respectively di-benzyl-(XI) or diphenyl-(XII)-urea. XI is formed also on heating IX with X to 180°. A solution of 3.76 g II in 40 ml absolute alcohol is mixed with 7.27 ml 16.4% alcoholic solution of HCl, there are added within 20 minutes with cooling with ice and stirring 3.88 g III, mixing is continued for 3 hours at ~20°, then the mixture is boiled for 4 hours, evaporated in vacuum and the residue is treated with 50% solution of K₂CO₃. Extracted with ether; the residue after evaporation of ether is heated 30 minutes in boiling water bath and ground with dry ether; yield of IV is 44.3%, MP 166-168°; hydrochloride MP 136-138° (from aqueous acetone). From mother liquor isolated 1 g VI, BP 87-89°/0.5 mm, 122-123°/14 mm, n²³D 1.4723, hydrochloride MP 300° (decomposition). Under analogous conditions were prepared V (BP 105-108°/0.6 mm, n¹⁶D 1.4587) and VII (n¹⁶D 1.4671). Mixture of 1 g IV and 10 ml HCl (1:1) boiled for 4 hours filtered, evaporated in vacuum, acetone is added, [REDACTED]

Card 2/3

USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61542

Abstract: NH_4Cl , filtrate evaporated, residue treated with 5% solution K_2CO_3 and extracted with CHCl_3 ; thus VII^a is obtained; picrate MP 158-160°.

Card 3/3

Mikhлина, Е. А.

✓ 14-Pyridylglutaric acid and products of its transformations. M. V. Rabinov, E. F. Mikhлина, and V. Ya. Furshatova (Institute of Applied Chemistry Research, Chem. Phys. Inst., Moscow). Zhur. Obshch. Khim. 26, 114-9 (1956); cf. C.A. 48, 12744b. — Heating 3 g. di-Et β -(4-pyridyl)glutarate with 23 ml. N KOH 3 hrs. gave on acidification and evapn. 09% β -(4-pyridyl)glutaric acid (I), m. 183-5° (from EtOH). I di-Et ester with N_2H_4 in EtOH gave 01.1% I dihydrazide, m. 180-0°. I di-Et ester and concd. $NH_4OH\cdot H_2O$ in 25 hrs. gave 84.8% I diamide, m. 192-4° (from EtOH). I di-Et ester (5 g.) refluxed 20 hrs. with 10.2 g. $NhCH_2NH_2$ gave 92.3% I bis(benzyldiamide), m. 163-3° (the use of $Et_2NCH_2CH_2NH_2$ similarly gave 57% I bis(diethylaminoethylamido) (II), m. 100-2°. Similarly was prep'd. I δ -(β -diethylaminopropylamido), a hygroscopic solid, b.p. 220-50°. II (3 g.) added to 1.04 g. LiAlH₄ in $Et_2O\cdot C_6H_6$, and refluxed 20 hrs., gave after wq. treatment, 58% I,6-di-(diethylaminoethylamino)-5-(4-pyridyl)pentane, b.p. 193-7°, n_D²⁰ 1.6147, whose picrate, yellow, m. 163-5°. Refluxing I di-Et ester with $Et_2NCH_2CH_2OH$, in which a little Na had been dissolved, with continuous distn. of EtOH 3 hrs. at 150° gave 64.3% I di-(diethylaminoethyl)ester, b.p. 193-7°. Reduction of I di-Et ester with LiAlH₄ in C_6H_6 (2 hrs. at reflux) gave 88% 3-(4-pyridyl)-1,5-dihydroxypentane, m. 64-6°. This (1.9 g.) acidified with alc. HCl, treated at EtOH by distn. *in vacuo*, soln. in dry $CHCl_3$, and treatment with 12 ml. SO_3 , gave after refluxing 1 hr. 42.7% 3-(4-pyridyl)-1,5-dichloropentane (III), b.p. 121-2°, after treatment with concd. K_2CO_3 . On standing this forms the quaternary salt, insol. in Et_2O . III (0.85 g.) refined with excess $PhCH_2NH_2$, 7 hrs. gave: N -benzyl-4-(4-pyridyl)piperidine, b.p. 160-4° (picrate, 1/2).

Robtsov, M.V., Mikhina, E.E., and Furshatova, V.Ya.

(m. 93-10°). Hydrogenation of 1 di-Et ester over Pt in alc. HCl gave 98% di-Et β -(4-piperidyl)glutarate; b_6 , 131-2°; nD₂₅ 1.4688, which with hot N NaOH in 8 hrs. gave the free acid, 93.8%, m. 213°, purified through the Ag salt, which is fused in H₂O/HCl salt, m. 174-5°. The di-Et ester conventionally treated gave: diamide, 51%, m. 180° (from Et-OH-Bt₂O); bis(benzylamide), 69%, m. 163-7°. (also prepd. from the acid HCl salt and SOCl₂, the crude acyl chloride then treated with RNH₂); bis(dithyldiaminoethylamide), 77%, b. & 170-90°. The bis(benzylamide) reduced with LiAlH₄ in Et₂O-dioxane to 33% 1,5-dibenzylamino-3-(4-piperidyl)pentane, b_6 , 235-70°. G.M. Kholostoff

USSR/ Organic Chemistry - Synthetic organic chemistry

E-2

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11:38

Author : Rubtsov M.V., Mikhlina Ye.Ye., Furshtatova V.Ya
Title : Preparation of Isonicotinic Acid

Orig Pub : Zh prokl. khimii, 1956, 29, No 6, 946-948

Abstract : A method has been developed for the preparation of isonicotinic acid (I) by oxidation with dilute HNO₃ of the mixture of α -methylolpicolines (II) formed on heating mixture of β - and γ -picolines (III, IV) with formalin (V) at atmospheric pressure. It is shown that in lieu of HNO₃, a mixture of HNO₃ and H₂SO₄ can be successfully utilized in the oxidation. An experimental study is made of the preparation of I from citric acid (VI); a more precise determination has been made of the conditions of preparation, with increased yields, of 2,6-dihydroxy isonicotinic acid (VII) and 2,6-dichlor isonicotinic acid (VIII); yields of I have been considerably increased. 117.6 g technical mixture III and IV (the mixture contains 15% water and 40% IV, on the dry basis) and 200 g V are boiled 15 hours; III and excess V are steam distilled, aqueous solution of II is concentrated to 160-180 ml and

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USSR/ Organic Chemistry - Synthetic organic chemistry

E-2

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11738

these are added, within 20 minutes, to 350 ml of 57.5% HNO₃, heated to 98°, heating is continued for 4 hours, after which neutralization is effected with 65-75 g Na₂CO₃ to obtain I, yield 77.5-85% (on basis of IV), MP 314°. From distillate, by addition of 32 g KOH and 48 g NaCl, are isolated 48-56 g III. Trimethyl ester of VI, 73 g, is shaken for 15-20 minutes with 730 ml 25% aqueous solution of NH₃, the mixture is evaporated in vacuum, 365 g of 73% H₂SO₄ are added, the mixture is slowly heated to 125°, and held at 125-130°.

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AUTHORS: Mikhлина, Ye. Ye., and Rubtsov, M. V.

TITLE: Synthesis of 3-Methylquinuclidine-2-carboxylic Acid (Sintez
3-metilkhinuklidinkarbonovoy kisloty)

PERIODICAL: Zhurnal Obshchey Khimii, 1957, Vol. 27, No. 1, pp. 77-83 (U.S.S.R.)

ABSTRACT: The synthesis of di-substituted quinuclidine derivatives - 3-methylquinuclidine-2-carboxylic acid from gamma-ethylpyridine, is described. Condensation of the gamma-ethylpyridine with dioxy-malonic ester and consequent conversion of 1,1-dicarboethoxy-2-(pyridyl-4')-propene-1 into 3-methylquinuclidine-2-carboxylic acid was considered the most simple synthesis method. However, instead of 1,1-dicarboethoxy-2-(pyridyl-4')-propene-1, ethyl ether alpha-oxy-alpha carboethoxy-beta-(pyridil-4)-butyric acid was obtained. Reduction of the diester in the presence of platinum oxide and consequent saponification and decarboxylation led to alpha-oxy-beta-(piperidyl-4)-butyric acid. In order to convert the latter compound into 3-methylquinuclidine-2-carboxylic acid, it was necessary to substitute the alpha-oxy group in the acid with a haloid. The acid was treated with thionyl chloride. At a temperature of 60-65° only acid chloride was formed; increased temperatures to 70-75° resulted in intensive resinification of

Card 1/2

Synthesis of β -Methylquinuclidine-2-carboxylic Acid

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the substance, and no change of the alpha-oxy group into Cl group was observed.

This result indicates that the alpha-oxy group in the acid is less active, as a result of which the synthesis of β -methylquinuclidine-2-carboxylic acid from alphaoxy-beta-(piperidyl-4)-butyric acid has proven impossible.

There are 5 references, of which 3 are Slavic.

ASSOCIATION: All-Union Scientific-Research Chemical-Pharmaceutical Institute im. S. Ordzhonikidze (Vsesoyuznyy Nauchno-Issledovatel'skiy Khimiko-Farmatseyticheskiy Institut im. S. Ordzhonikidze).

PRESENTED BY:

SUBMITTED: January 30, 1956

AVAILABLE:

Card 2/2

MIKHLINA, E.E.

*✓ Synthesis of the lactone of 3-hydroxymethyl-3-carboxy-
methylquinuclidine.* B. P. Mikhlina and M. V. Rubtsov
(S. Ordzhonikidze All Union Chem. Pharm. Research
Inst., Moscow). Zhur. Obshchey Khim. 27, 691-8 (1957); cf.
C.A. 50, 3471.—Refluxing 84.4 g. Bt 3-(4-pyridyl)propen-
ate with 81 g. formalin in 370 ml. H₂O 1 day, steam distg.
the residual CH₂O, and evapn. the residue gave 69.5%
4-C₆H₄N(C(=O)CH₂OH).CH₂.CO.O.CH₂, m. 132-4° (from dry
EtOH); with concd. HCl it gave the HCl salt, m. 218-17°
(from 80% EtOH). The latter shaken with Pt catalyst
and H in dry EtOH gave 65% piperidyl analog, isolated as
the HCl salt (I), decomp. 211-13° (from EtOH-Et₂O), also
isolated in 28.5% yield on similar hydrogenation in 4%
HCl; the free base m. 141-3° (from CHCl₃). The HCl
salt (15 g.) heated 4 hrs. at 60-5° with 60 ml. SOCl₂ in
C₆H₆ and 4 hrs. at 70-8° gave, after reprecip. of the solid, wash-
ing it with C₆H₆ and Et₂O, volat. in warm EtOH and cooling,
79.5% 4-C₆H₄NCH₂C(=O)CH₂Cl.CH₂.CO.O.CH₂.HCl. (II),
decomp. 225-7°, which shaken with Ag₂O in H₂O 8 hrs.
gave colorless Ag₂OUC₆H₄C(=O)CH₂CH(C₆H₄N-4), which
taken up in aq. HCl and filtered gave, on evapn., I, identical
with the above. Treating 7.03 g. II in 200 ml. EtOH
with 68.7 ml. 0.5N alc. KOH, filtering, evapg., and re-
fluxing the residue with 70 ml. dry pyridine 8 hrs. and
cooling gave 69.8% 3-hydroxymethyl-3-carboxymethylquinucli-
dine lactone-HCl, decomp. 210-12° (from dry EtOH); free
lactone, b.p. 149°, m. 137-9°; picrate, m. 234-6° (from 80%
EtOH). The lactone heated with N₂H₄.H₂O in EtOH 10
hrs. gave 98% 3-(hydroxymethyl)quinuclidyl-3-acetic acid
hydrazone, m. 148-8° (from CHCl₃-Et₂O). Similarly was
prepd. the phenylhydrazone, decomp. 181-3° (from EtOH-
Et₂O).

G. M. Kusolapoff

Mikhailina, Ye. Ye.

AUTHORS: Mikhailina, Ye. Ye., Rubtsov, M. V. 70-1-22/21

TITLE: The Synthesis of Quinuclidine-3-Acetic Acid (Sintez khinuklidin-3-oksusnoy kisloty)

PERIODICALS: Zhurnal Obshchey Khimii, '958, Vol.28, Nr 1, pp 115-116 (SSSR)

ABSTRACT: The present paper describes the synthesis of quinuclidine-3-acetic acid which makes it possible to transfer the investigation to the 3-substituted derivatives of quinuclidine. 4-(β -oxyethyl) was first used as initial product for this synthesis with the attempt of synthesizing 1,1,1-trichloro-2-oxy-3-(pyridyl-4')-4-oxybutane from this product by reaction with chloral. This attempt failed, as only resin like products were obtained. The tests with 4-(β -acetoxyethyl)-pyridine (83%) obtained from 4-(β -oxyethyl) pyridine (formula I) only yielded 4-vinylpyridine as final product. 4-(β -methoxyethyl)-pyridine (VI) proved to be a more stable compound during the influence of chloral. The condensation of this product with chloral in the presence of acetic piperidine (see the reaction scheme) led to 1,1,1-trichloro-2-

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The Synthesis of Quinuclidine-3-Acetic Acid

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-3-(pyridyl-4')-4-methoxybutane (VII), where no substituted 4-vinylpyridine was produced (as it was the case in the analogous reaction of 4-(β -acetoxymethyl)-pyridine with chloroacetyl). The compound (VII) was by reaction with potassium alkoholate converted to 4-methoxy-3-(pyridyl-4')-propanoic acid (VIII). Its ethyl ester (IX) was at room temperature and in the presence of a platinum catalyst converted to the ethyl ester of methoxy-3-(piperidyl-4')-butyric acid (X). In order to exchange the methoxy group for halide and for the purpose of synthesis of the derivative of quinuclidine the compound (X) was heated at 100 - 120°C with 67% hydrobromic acid in a tube soldered shut. 4-bromo-3-(piperidyl-4') butyric acid synthesized in this connection was esterified and in the presence of pyridine subjected to cyclization. The ethyl ester of quinuclidine-3-acetic acid (XI) finally resulted which was after saponification converted to quinuclidine-3-acetic acid. There are 3 references 4 of which are Slav... .

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The Synthesis of Quinuclidine-3-Acetic Acid

70-1-22/03

ASSOCIATION: All-Union Scientific Chemical-Pharmaceutical Research Institute imeni Ordzhonikidze
(Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut im. S. Ordzhonikidze)

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Card 3/3

1. Chemistry 2. Quinuclidine-Synthesis 3. Acetic acid

AUTHORS: Furshatova, V. Ya., Mikhлина, Ye. Ye., 79-28-5-8/69
Rubtsov, M. V.

TITLE: Synthesis of 6,7-Di-substituted 1-Azabicyclo-(3,2,1)-Octane (Sintez 6,7-dizamesnchennykh 1-azabitsiklo-(3,2,1)-oktana)

PERIODICAL: Zhurnal Obshchey Khimii, 1958, Vol. 28, Nr 5,
pp. 1170-1176 (USSR)

ABSTRACT: In the last publication by the authors (Reference 1) a simple synthesis of 6-carboxymethyl-1-azabicyclo-(3,2,1)-octane-7-carboxylic acid and of its derivatives was described. Most interesting of these compounds were the properties of the ethyl esters of 6-carbethoxymethyl-1-azabicyclo-(3,2,1)-octane-7-carboxylic acid. Thus this ester hydrolyzed easily in aqueous solution under formation of an acidic ester. The same way also reacts the isomeric ethyl ester of 3-carbethoxymethylquinuclidine-2-carboxylic acid which converts to 3-carbethoxymethylquinuclidine on the same conditions. The comparison of the two isomeric diesters makes possible the assumption, that the saponification of

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Synthesis of 6,7-Di-substituted 1-Azabicyclo-(3,2,1)-Octane

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of the carbethoxyl group in ethyl ester of the 6-*aryloxy*-
xymethyl-1-azabicyclo-(3,2,1)-octane-7-carboxylic acid
(in the mentioned scheme) takes place in position 7 and
that the acidous ester forming on this occasion has the
structure (II) of the scheme. From this a whole number
of 7-alkyl-(aryl)-aminoethyl-6-(β -oxyethyl)-1-azabicyclo-
-octanes and of esters of 7-dialkylaminoethyl-6-(β -oxyethyl)
-1-azabicyclo-(3,2,1)-octane were obtained. The compound
(II) converts to compound (III) by means of thionylchloride;
this compound was further treated with alkyl-(aryl)
amines. The amides (IV) obtained then were reduced to the
compound (V) by means of lithium aluminum hydride. On treat-
ing this with chlorine anhydrides or some acids the cor-
responding esters (VI) resulted. On the conversion of
(V zh - ~~etc~~) with thionylchloride the compound (VII) was
obtained in which the chlorine atom in the 6- β -chlorethyl
group is of little activity as far as reactivity is concerned.
There are 2 Soviet references.

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